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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.	
10/765,929	01/29/2004	Toshiaki Aono	Q79636	2659	
23373	7590 12/27/2005		EXAM	EXAMINER	
SUGHRUE MION, PLLC			MARTIN, LAURA E		
SUITE 800	YLVANIA AVENUE, N.W.		ART UNIT	ART UNIT PAPER NUMBER	
WASHINGTON, DC 20037			2853		
			DATE MAILED: 12/27/2009	DATE MAILED: 12/27/2005	

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)			
	10/765,929	AONO ET AL.	(M)		
Office Action Summary	Examiner	Art Unit	1		
	Laura E. Martin	2853			
The MAILING DATE of this communication app Period for Reply	pears on the cover sheet with the c	orrespondence ad	dress		
A SHORTENED STATUTORY PERIOD FOR REPL' WHICHEVER IS LONGER, FROM THE MAILING DA - Extensions of time may be available under the provisions of 37 CFR 1.1 after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period of - Failure to reply within the set or extended period for reply will, by statute Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION 36(a). In no event, however, may a reply be tim will apply and will expire SIX (6) MONTHS from , cause the application to become ABANDONE	N. nely filed the mailing date of this or D (35 U.S.C. § 133).			
Status					
1) Responsive to communication(s) filed on 29 Ja	anuary 2005.				
2a) ☐ This action is FINAL . 2b) ☑ This	action is non-final.				
3) Since this application is in condition for allowance except for formal matters, prosecution as to the merit					
closed in accordance with the practice under E	Ex parte Quayle, 1935 C.D. 11, 45	53 O.G. 213.			
Disposition of Claims					
4) Claim(s) 1-34 is/are pending in the application. 4a) Of the above claim(s) is/are withdraw 5) Claim(s) is/are allowed. 6) Claim(s) 1-34 is/are rejected. 7) Claim(s) is/are objected to. 8) Claim(s) are subject to restriction and/o	wn from consideration.				
Application Papers	•				
9) The specification is objected to by the Examine 10) The drawing(s) filed on is/are: a) acc Applicant may not request that any objection to the Replacement drawing sheet(s) including the correct 11) The oath or declaration is objected to by the Examine	epted or b) objected to by the I drawing(s) be held in abeyance. See tion is required if the drawing(s) is obj	e 37 CFR 1.85(a). jected to. See 37 CF	, ,		
Priority under 35 U.S.C. § 119	•				
 12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of: 1. Certified copies of the priority document 2. Certified copies of the priority document 3. Copies of the certified copies of the priority application from the International Bureau * See the attached detailed Office action for a list 	s have been received. s have been received in Applicati rity documents have been receive u (PCT Rule 17.2(a)).	on No ed in this National	Stage		
Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08) Paper No(s)/Mail Date 1/29/04.	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal P 6) Other:	ate	O-152)		

DETAILED ACTION

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-5, 9-19, 21-24 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kubodera (JP 10095942) in view of Ishizuka et al. (JP 2001181548) and Kimura et al. (US 6521031).

Kubodera teaches an inkjet recording ink and image forming method [0001], comprising a pigment and a compound represented by the following general formula (I): $R-X-(Y)_n-H$, wherein the general formula (I), R represents a hydrophobic group [0005], or a group derived from a hydrophobic polymer; n is an integer from 10 to 3500 [0005]; and structural units of repeated Y comprise at least one structural unit represented by A, C, or D [0005], and further comprise 0-40% by mole of structural units represented by B [0045]:

$$A : -(CH_{2}-C) - CH_{2} - C$$

wherein in structural units A through D, R¹ represents a hydrogen atom or an alkyl group having 1 to 6 carbon atoms; R² represents a hydrogen atom or an alkyl group having 1 to 10 carbon atoms; R³ represents a hydrogen atom or a methyl group; R⁴ represents a hydrogen atom, -CH₃, -CH₂COOH, or an ammonium salt thereof or an alkali metal salt thereof or -CN; Z1 (X) represents a hydrogen atom, -COOH, or an ammonium salt thereof or alkali metal salt thereof, or -CONH₂; and Z² (Y) represents -COOH or an ammonium salt thereof or alkali metal salt thereof, SO3H or an ammonium salt thereof or alkali metal salt thereof, -OSO₃H or an ammonium salt thereof or alkali metal salt thereof, -CH₂SO₃H or an ammonium salt thereof or alkali metal thereof, -CONHC(CH₃)₂CH₂SO₃H or an ammonium salt thereof or alkali metal salt thereof, or – CONHCH₂CH₂CH₂N⁺(CH₃)₃Cl⁻[0007]. Kubodera also teaches the hydrophobic group represented by R in general formula (I) is an aliphatic group or an aromatic group, alicyclic group, is selected from the group consisting of alkyl, alkenyl, alkynyl, phenyl, and naphthyl groups [0030]. Kubodera also teaches the hydrophobic group represented by R in general formula (I) is an alkyl group having 3 to 70 carbon atoms [0031] wherein polymerization degree of R in the general formula (I) is from 2 to 500 [0032]: R is a group derived from at least one hydrophobic polymer selected from the group consisting of polystyrene, polymethacrylic acid ester, polyacrylic acid ester, polyvinyl chloride, and derivatives thereof [0032]. Kubodera also teaches the structural unit A is a structural unit derived from vinyl alcohol, α-methylvinyl alcohol or αpropylvinyl alcohol [0043]; the structural unit B is a structural unit derived from vinyl acetate, vinyl formate, vinyl propionate, or an α-substitution product thereof [0043]; the

structural unit C is a structural unit derived from acrylic acid, methacrylic acid, itaconic acid, maleic acid, an ammonium salt thereof or a metal salt thereof [0043]. Kubodera also teaches a mass ratio of R to (Y)_n in general formula (I) is from 0.01 to 2, the mass ratio being calculated using atomic weights of respective atoms in R to (Y)_n [0048]; (Y)_n comprises, as structural units thereof, ethylene, propylene, isobutene, acrylonitrile, acrylamide, methacrylamide, N-vinylpyrrolidone, vinyl chloride, or vinyl fluoride [0046]. Kubodera also teaches the ink further comprising water [0073].

Kubodera does not teach X representing a bivalent linking group having a hetero bond or the hetero bond in general formula (I) selected from the group consisting of an ether bond, an ester bond, a thioether bond, a thioester bond, a sulfonyl bond, an amide bond, an imide bond, a sulfonamide bond, a urethane bond, a urea bond, and a thiourea bond. Kubodera also does not teach colored fine particles including an oilsoluble dye and an oil-soluble polymer, wherein the oil soluble polymer has a dissociable group, begin at least one of a carboxyl group or a sulfonic acid group, in an amount of 0.2 to 4.0 mmol/g, the colored fine particles being prepared by adding to an aqueous phase an organic phase including the oil-soluble dye, selected from the group consisting of an anthraquinone dye, naphthoquinone dye, styryl dye, indoaniline dye, azo dye, nitro dye, coumarin dye methane dye, porphyrin dye, azaporphyrin dye, and phthalocyanine dye, and the oil-soluble polymer, and emulsifying and dispersing a resultant mixture, and the ink having a viscosity of 30 mPa*s.

Kimura et al. teaches X representing a bivalent linking group having a hetero bond or the hetero bond in general formula (I) selected from the group consisting of an ether

bond, an ester bond, a thioether bond, a thioester bond, a sulfonyl bond, an amide bond, an imide bond, a sulfonamide bond, a urethane bond, a urea bond, and a thiourea bond (C6, L15-26).

It would have been obvious to one of ordinary skill in the art at the time of the invention to modify the invention of Kubodera with the disclosure of Kimura et al. in order to provide for a stable ink composition.

Ishizuka et al. teaches colored fine particles including an oil-soluble dye and an oil-soluble polymer (abstract), wherein the oil soluble polymer has a dissociable group, begin at least one of a carboxyl group or a sulfonic acid group, in an amount of 0.2 to 4.0 mmol/g [0011], the colored fine particles being prepared by adding to an aqueous phase an organic phase including the oil-soluble dye [0059], selected from the group consisting of an anthraquinone dye, naphthoquinone dye, styryl dye, indoaniline dye, azo dye, nitro dye, coumarin dye methane dye, porphyrin dye, azaporphyrin dye, and phthalocyanine dye [0008-0009], and the oil-soluble polymer, and emulsifying and dispersing a resultant mixture [0059], and the ink having a viscosity of 30 mPa*s [0086].

It would have been obvious to one of ordinary skill in the art at the time of the invention to modify the invention of Kubodera with the disclosure of Ishizuka et al. in order to create a more stable ink for printing.

Claims 6 and 7 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kubodera (JP 10095942), Ishizuka et al. (JP 2001181548) and Kimura et al. (US 6521031) in further view of Yabuki et al. (US 20020067399).

Kubodera, Ishizuka et al., and Kimura et al. teach the ink of claim 1; however, none disclose the ink wherein the oil soluble dye is contained in an amount of 0.5-50% by mass based on a total mass of the ink or wherein the oil soluble polymer is contained in an amount of 10-500% by mass based on a mass of the oil soluble dye.

Yabuki et al. teaches the ink wherein the oil soluble dye is contained in an amount of 0.5-50% by mass based on a total mass of the ink [0070] or wherein the oil soluble polymer is contained in an amount of 10-500% by mass based on a mass of the oil soluble dye [0199].

It would have been obvious to one of ordinary skill in the art at the time of the invention to modify the disclosures of Kubodera, Ishizuka et al., and Kimura et al. with that of Yabuki et al. in order to create a more stable ink composition.

Claim 8 is rejected under 35 U.S.C. 103(a) as being unpatentable over Kubodera (JP 10095942), Ishizuka et al. (JP 2001181548) and Kimura et al. (US 6521031) in further view of Ishizuka et al. (US 2002088294).

Kubodera, Ishizuka et al. (548), and Kimura et al. teach the ink of claim 1; however, none disclose the compound represented by formula (I) contained in an amount of 1 to 50 % by mass based on a mass of colored fine particles.

Ishizuka et al. (294) teaches the compound represented by formula (I) contained in an amount of 1 to 50 % by mass based on a mass of colored fine particles (abstract).

It would have been obvious to one of ordinary skill in the art at the time of the invention to modify the disclosures of Kubodera, Ishizuka et al. (548), and Kimura et al. with that of Ishizuka et al. (294) in order to create a more stable ink composition.

Claim 20 is rejected under 35 U.S.C. 103(a) as being unpatentable over Kubodera (JP 10095942), Ishizuka et al. (JP 2001181548) and Kimura et al. (US 6521031) in further view of Leppard et al. (US 60480660).

Kubodera, Ishizuka et al. and Kimura teach the ink of claim 1; however, none of these references teach a structural unit D selected from the group consisting of – CH₂CH(OH)CH₂O-, -CH₂C(CH₃)(OH)CH₂O-, and -CH₂C(C₂H₅)(OH)CH₂O-.

Leppard et al. teaches a structural unit D selected from the group consisting of – CH₂CH(OH)CH₂O-, -CH₂C(CH₃)(OH)CH₂O-, and -CH₂C(C₂H₅)(OH)CH₂O- (C2, L54).

It would have been obvious to one of ordinary skill in the art at the time of the invention to combine the invention of claim 1 with the disclosure of Leppard et al. in order to provide for a stable ink composition.

Claims 25, 27, 29 and 34 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kubodera (JP 10095942) in view of Ishizuka et al. (JP 2001181548) and Ito et al. (US 6509125).

Kubodera teaches an inkjet recording ink and image forming method [0001], comprising a pigment and a compound represented by the following general formula (I): R-X-(Y)_n-H, wherein the general formula (I), R represents a hydrophobic group [0005], or a group derived from a hydrophobic polymer; n is an integer from 10 to 3500 [0005]: and structural units of repeated Y comprise at least one structural unit represented by A, C, or D [0005], and further comprise 0-40% by mole of structural units represented by B [0045]:

A:
$$-(CH_2-C)$$
 B: $-(CH_2-C)$ OH

C: $-(CH_2-C)$ D: $-(CH_2-C)$ OH

wherein in structural units A through D, R¹ represents a hydrogen atom or an alkyl group having 1 to 6 carbon atoms; R² represents a hydrogen atom or an alkyl group having 1 to 10 carbon atoms; R³ represents a hydrogen atom or a methyl group; R⁴ represents a hydrogen atom, -CH₃, -CH₂COOH, or an ammonium salt thereof or an alkali metal salt thereof or -CN; Z1 (X) represents a hydrogen atom, -COOH, or an ammonium salt thereof or alkali metal salt thereof, or -CONH₂; and Z² (Y) represents -COOH or an ammonium salt thereof or alkali metal salt thereof, SO3H or an ammonium salt thereof or alkali metal salt thereof, -OSO₃H or an ammonium salt thereof or alkali metal salt thereof, -CH₂SO₃H or an ammonium salt thereof or alkali metal thereof, -

CONHC(CH₃)₂CH₂SO₃H or an ammonium salt thereof or alkali metal salt thereof, or – CONHCH₂CH₂CH₂N⁺(CH₃)₃Cl⁻ [0007]. Kubodera also teaches the hydrophobic group represented by R in general formula (I) is an aliphatic group, aromatic group, or an alicyclic group [0030] and R is a group derived from at least one hydrophobic polymer selected from the group consisting of polystyrene, polymethacrylic acid ester, polyacrylic acid ester, polyvinyl chloride, and derivatives thereof [0032].

Kubodera does not teach X representing a divalent linking group having a hetero bond or the hetero bond in general formula (I) Kubodera also does not teach colored fine particles including an oil-soluble dye and an photopolymerizable monomer.

Kimura et al. teaches X representing a bivalent linking group having a hetero bond or the hetero bond in general formula (I) selected from the group consisting of an ether bond, an ester bond, a thioether bond, a thioester bond, a sulfonyl bond, an amide bond, an imide bond, a sulfonamide bond, a urethane bond, a urea bond, and a thiourea bond (C6, L15-26).

It would have been obvious to one of ordinary skill in the art at the time of the invention to modify the invention of Kubodera with the disclosure of Kimura et al. in order to provide for a stable ink composition.

Ito et al. teaches colored fine particles including an oil-soluble dye (C2, L25-36) and a photopolymerizable monomer (C13, L48-60).

It would have been obvious to one of ordinary skill in the art at the time of the invention to modify the invention of Kubodera with the disclosure of Ishizuka et al. in order to create a more stable ink for printing.

Claims 26, 29, and 33 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kubodera (JP 10095942), Ito et al. (US 6509152) and Kimura et al. (US 6521031) in further view of Ishizuka et al. (US 2001181548).

Kubodera, Ito et al., and Kimura et al. teach the ink of claim 1; however, none teach the colored fine particles are prepared by adding to an aqueous phase an organic phase including oil-soluble dye, and emulsifying and dispersing a resultant mixture; wherein the oil-soluble dye is selected from the group consisting of an anthraquinone dye, naphthoquinone dye, styryl dye, indoaniline dye, azo dye, nitro dye, coumarin dye methane dye, porphyrin dye, azaporphyrin dye, and phthalocyanine dye, and the oil-soluble polymer, and emulsifying and dispersing a resultant mixture, and the ink having a viscosity of 30 mPa*s.

Ishizuka et al. teaches the colored fine particles are prepared by adding to an aqueous phase an organic phase including oil-soluble dye, and emulsifying and dispersing a resultant mixture [0059]; wherein the oil-soluble dye is selected from the group consisting of an anthraquinone dye, naphthoquinone dye, styryl dye, indoaniline dye, azo dye, nitro dye, coumarin dye methane dye, porphyrin dye, azaporphyrin dye, and phthalocyanine dye [0008-0009], and the oil-soluble polymer, and emulsifying and dispersing a resultant mixture, and the ink having a viscosity of 30 mPa*s [0086].

It would have been obvious to one of ordinary skill in the art at the time of the invention to modify the invention of claim 1 with the disclosure of Ishizuka et al. in order to provide a more stable ink composition.

Claims 30 and 32 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kubodera (JP 10095942), Ito et al. (JP 6509125) and Kimura et al. (US 6521031) in further view of Yabuki et al. (US 20020067399).

Kubodera, Ito et al., and Kimura et al. teach the ink of claim 1; however, none disclose the ink wherein the oil soluble dye is contained in an amount of 0.5-50% by mass based on a total mass of the ink or wherein the oil soluble polymer is contained in an amount of 10-500% by mass based on a mass of the oil soluble dye.

Yabuki et al. teaches the ink wherein the oil soluble dye is contained in an amount of 0.5-50% by mass based on a total mass of the ink [0070] or wherein the oil soluble polymer is contained in an amount of 10-500% by mass based on a mass of the oil soluble dye [0199].

It would have been obvious to one of ordinary skill in the art at the time of the invention to modify the disclosures of Kubodera, Ito et al., and Kimura et al. with that of Yabuki et al. in order to create a more stable ink composition.

Claim 31 is rejected under 35 U.S.C. 103(a) as being unpatentable over Kubodera (JP 10095942), Ito et al. (JP 6509125) and Kimura et al. (US 6521031) in further view of Ishizuka et al. (US 2002088294).

Kubodera, Ito et al., and Kimura et al. teach the ink of claim 1; however, none disclose the compound represented by formula (I) contained in an amount of 1 to 50 % by mass based on a mass of colored fine particles.

Ishizuka et al. teaches the compound represented by formula (I) contained in an amount of 1 to 50 % by mass based on a mass of colored fine particles (abstract).

It would have been obvious to one of ordinary skill in the art at the time of the invention to modify the disclosures of Kubodera, Ito et al., and Kimura et al. with that of Ishizuka et al. in order to create a more stable ink composition.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Laura E. Martin whose telephone number is (571) 272-2160. The examiner can normally be reached on Monday - Friday, 7:00 - 3:30.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Stephen D. Meier can be reached on (571) 272-2149. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Laura E. Martin

MANISH S. SHAH